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18 States of Matter **Three**
States of Matter - Solids,
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You are already familiar
with the three common states
of matter: solid, liquid,

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and gas. Solid objects litter the room around you. For example, you can easily recognize the shape of your desk; you know that your backpack cannot hold seven textbooks. You encounter liquids throughout the day

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Chapter 13: States of Matter

Chapter 13 States of
Matter137 SECTION 13.1 THE
NATURE OF GASES (pages
385–389) This section
introduces the kinetic

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theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386) 1.

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*Name Date Class STATES OF
MATTER 13*

There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can

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GuideAnswers refer you to somebody.)

Those three states are solid, liquid, and gas.

These three states are quite different. The main difference is in their particles.

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*Chapter 13: States of Matter
- Chemistry by Anna*

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theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance.

Chapter 13 States Of Matter

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137 SECTION 13.1 THE NATURE OF GASES (pages 385–389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains

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how temperature is related to the kinetic energy of the particles of a substance.

*Chapter 13 States Of Matter
Worksheet*

Chapter 13: States of
Matter. STUDY. PLAY. Kinetic

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Molecular Theory. Explains the properties of gases in terms of the energy, size, and motion of their particles. **Elastic Collision.** Describes a collision in which kinetic energy may be transferred

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between colliding particles but the total kinetic energy of the two particles remains the same.

*Chapter 13: States of Matter
Flashcards | Quizlet
Chemistry (12th Edition)*

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Answers to Chapter 13 -
States of Matter - 13.1 The
Nature of Gases - 13.1
Lesson Check - Page 424 8
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0132525763, ISBN-13:

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Chapter 13 - States of

Matter ...

all matter consists of tiny

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particles that are constantly in motion What are the three assumptions of the kinetic theory as it applies to gases? -The particles in a gas are considered to be small, hard spheres with an

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insignificant volume. -The motion of the particles in a gas are rapid, constant, and random.

*Chapter 13: States of Matter
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The Sustainable Development

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Goals are a call for action by all countries – poor, rich and middle-income – to promote prosperity while protecting the planet. They recognize that ending poverty ...

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13.1 The Nature of Gases -
Sample Problem 13.1; 13.1

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Guides Answers

The Nature of Gases - 13.1
Lesson Check; 13.2 The
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Chemistry & You; 13.2 The
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Lesson Check; 13.3 The
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Solids - 13.3 Lesson Check;
13.4 Changes of State -
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The Nature of Gases - 13.1
Lesson Check; 13.2 The
Nature of Liquids -
Chemistry & You; 13.2 The

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Nature of Liquids - 13.2

Lesson Check; 13.3 The

Nature of Solids - Chemistry

& You; 13.3 The Nature of

Solids - 13.3 Lesson Check

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Chapter 13 - States of

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Title: Chapter 13 States of Matter 1 Chapter 13 States of Matter 2 Kinetic Theory as Applied to Gases
Fundamental assumptions about gases. The particles in a gas are considered to

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be small, hard spheres with an insignificant volume. Between particles in a gas there is empty space. No attractive or repulsive forces exist between the particles. 3

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Chapter 13 States Of Matter Worksheet

The attacks the USPS continues to face are not just attacks on the postal service but attacks on Black lives. To defund the USPS would be to deny future

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and dishonor the legacy of
Black postal workers. Now,
we're taking this matter
into our own hands by
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