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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. a) an ionic compound that is quite soluble in water (a) HCl c) an ionic compound that is not very soluble in water (b) NaNO_3

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Chemistry Chapter 13 review. absolute zero. universal gas constant. Boyle's law. Charles's law. a temperature at which a gas theoretically occupies zero volume. $0.08206 \text{ L atm/mol K}$. at constant temperature, the volume of a given amount of gas is directly proportional to the absolute temperature. at constant pressure, the volume of a given amount of gas is directly proportional to the absolute temperature.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a) The notation for sodium chloride, NaCl , stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d) In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

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6 Chemical Bonding

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

Assessment Chapter Test A - Ed W. Clark High School

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